

0.5 Scientific Notation Pretest/Checkup

1. What is 0.006 70 m in correct scientific notation?

- a) 6.70×10^{-3} m
- b) 6.70×10^{-3}
- c) 6.70×10^3
- d) 6.70×10^3 m
- e) none of the above

2. What is 1.59×10^7 g in conventional notation?

- a) 0.000 000 159 g
- b) 1 590 000 000 g
- c) 15 900 000 g
- d) 0.000 000 159
- e) 1 590 000 000

3. What is 4.1×10^{-5} ns in conventional notation?

- a) 0.000 041 ns
- b) 0.000 004 1 ns
- c) 410 000 ns
- d) 4 100 000 ns
- e) none of the above

4. What is 350 000 mol in correct scientific notation?

- a) 3.5×10^{-5} mol
- b) 3.5×10^{-4} mol
- c) 3.5×10^4 mol
- d) 3.5×10^5 mol
- e) none of the above

5. When using correct scientific notation, the base number must fall between ...

- a) 0 and 1
- b) 1.000 and 9.999 (repeating decimals)
- c) 1 and 100
- d) 0 and 100
- e) none of the above

6. In correct scientific notation, the number ten thousand would be ...

- a) 1×10^3
- b) 1×10^4
- c) 1×10^5
- d) 1×10^{-5}
- e) 1×10^{-4}

7. In correct scientific notation, the number 1/1000 would be ...

- a) 1×10^4
- b) 1×10^3
- c) 1×10^{-4}
- d) 1×10^{-2}
- e) 1×10^{-3}

8. In correct scientific notation, the number one-millionth would be ...

- a) 1×10^{-6}
- b) 1×10^{-7}
- c) 1×10^{-5}
- d) 1×10^5
- e) 1×10^6

9. What is 5.6×10^{-7} s + 3.65×10^{-6} s?
DO THIS WITHOUT A CALCULATOR!

- a) 9.25×10^{-13} s
- b) 9.25 s
- c) 9.25×10^{-7} s
- d) 4.21×10^{-6} s
- e) none of the above

10. What is 6.72×10^5 g - 3.65×10^3 g?
DO THIS WITHOUT A CALCULATOR!

- a) 668 350 g
- b) 3.07×10^2 g
- c) 6.68×10^2 g
- d) 3.07×10^8 g
- e) none of the above

Answer Key

1. (a)
2. (c)
3. (a)
4. (d)
5. (b)
6. (b)
7. (e)
8. (a)
9. (d)
10. (a)